M.V.MUTHIAH GOVERNMENT ARTS COLLEGE FOR WOMEN

DINDIGUL – 624 001, TAMIL NADU, INDIA

(AFFILIATED TO MOTHER TERESA WOMEN'S UNIVERSITY, KODAIKANAL)

RE-ACCREDITED WITH NAAC 'A' GRADE

DEPARTMENT OF CHEMISTRY

ONLINE INTERACTIONS - ACTIVITY REPORT

Event Name: National Level E-Quiz on Chemistry

Start Date: 3.06.2020

End Date: 4.06.2020

No. of Participants: 942

"National Level E-Quiz on Chemistry" was conducted by

the Department of Chemistry, M.V.Muthiah Government Arts

College for Women, Dindigul on 3.06.2020 & 4.06.2020.

942 participants of various departments from various

Institutions of all over India had taken part in the E-Quiz. 657

Participants had successfully completed and received their

F-Certificates.

This Quiz competition contains 20 Multiple Choice Based

questions about General Chemistry.

The objective of conducting this E-quiz competition is to

kindle the knowledge of basic science among the students,

scholars and faculties and to enrich their basic knowledge of General Chemistry during this Corona Pandemic Lockdown situation.

E-Certificates were sent through registered E-mail id for the participants who have secured 60% of marks and above in the Quiz.

No. of Participants attended: 942

No. of Participants Secured 60% of marks and above: 657

Link for the E-Quiz: https://forms.gle/8duo4e8yjyeNqZZF8



National Level E-Quiz on Chemistry

M.V.Muthiah Government Arts College, Dindigul, Tamil Nadu Date: 03/06/2020 & 04/06/2020 Time :10.00am to 6.00pm forms.gle

Greetings to All,

PG Department of Chemistry,

M.V.Muthiah Government Arts College for Women , Dindigul, happily invites you to take part in National Level E-Quiz on Chemistry .

Date:03/06/2020 & 04/06/2020 Time:10.00am to 6.00 pm

No Registration Fee

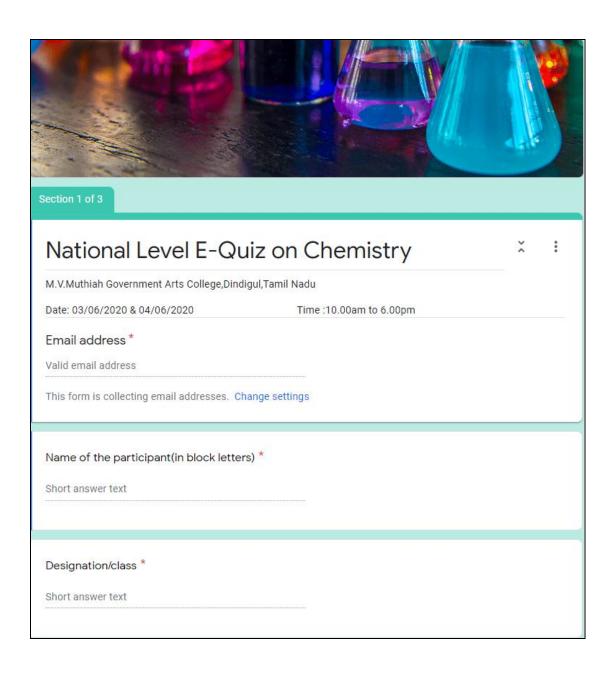
E-Certificates will be provided to the participants who scores 60% in the quiz.

With Regards, Dr.C.Ladha Pooranam Principal.

Dr.M.Dhanalakshmi, Convenor and Head , Department of Chemistry.

Here is the link for the Quiz: https://forms.gle/8duo4e8yjyeNqZZF8

2:34 pm //



Name of the institution *
Short answer text
Department *
Short answer text
After section 1 Continue to next section
Section 2 of 3
National Level E-Quiz on Chemistry × :
Description (optional)
1. Who discovered that water is composed of two parts hydrogen and one part oxygen by volume?
Pierre Robiquet
Joseph Louis Gay-Lussac
Henri-Victor Regnault
○ Georges Urbain

2. Which among the following would cause the bright red color due to bursting of crackers?
Strontium
Sodium
Sulphur
○ Magnesium
3. Presence of which among the following minerals in Banana makes them slightly radioactive?
Sodium
Calcium
Magnesium
OPotassium
4. Which among the following chemicals is used in Photography?
O Silver Bromide
O Potassium Chloride
O Sodium Bromide
Magnesium Chloride

5. Which among the following is called "Fool's Gold"? Copper Sulphate Iron Sulfide Brass Silver Bromide
6.LPG is a mixture of Methane and butane Butane and propane Methane and propane Ethane and propane
7.Who coined the word 'nanotechnology'? Circ Drexler Richard Feynmann Sumio Tijima Richard Smalley

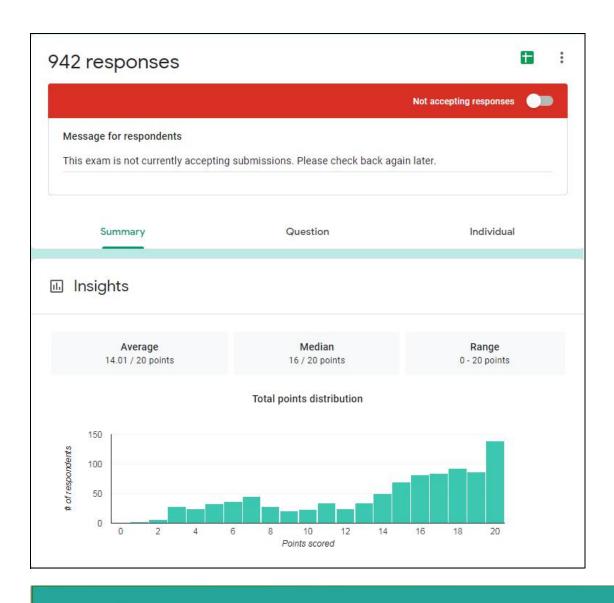
8. Sp3d2 hybridization is present in [Co(NH3)63+], find its geometry
Square planar geometry
octahedral geometry
tetragonal geometry
tetrahedral geometry
 Chlorination is a process used for water purification The disinfecting action of chlorine is mainly due to.
The formation of hydrochloric acid when chlorine is added to water.
The formation of hypochlorous acid when chlorine is added to water.
The formation of nascent oxygen when chlorine is added to water.
The formation of hydrogen when chlorine is added to water.
10.Formic acid is produced by
White Ants
○ Cockroaches
○ Red ants
Mosquitoes

Section 3 of 3		
National Level E-Quiz on Chemistry Description (optional)	×	:
11.Trans boundary Pollution or Acid rain is caused by Carbon dioxide Hydrocarbon Nitrogen oxide and sulphur dioxide Carbon monoxide		
12.Stains of rust on clothes can be removed by Alcohol Petrol Hydrogen peroxide Oxalic acid		

13.The National Chemical Laboratory is situated in
New Delhi
O Bangalore
O Pune
O Patna
14.What is the molecular geometry of a Methane molecule?
○ Linear
O Square Planar
○ Tetrahedral
Trigonol bipyramidal
15.What is the chemical formula of Rust?
○ Fe203. nH20
Fe2Cl3. nH2O
FeCO2. nH2O
FeO. nH2O

16. Which of the following is a polymer of amino acids?
Starch
Cellulose
O Proteins
None of the above
17.Cobalt - 60 is commonly used in radiation therapy because it emits
O Alpha - rays
O Beta rays
○ Gamma rays
○ X - rays
18.What do we call the region of space around the nucleus where the probability of finding an electron is maximum?
Shell
○ sub-shell
Orbital
electron cloud

19.Which of the following is true about a water molecule?
It has a bent-shaped structure
○ The bond angle is 120°
It is a non-polar solvent
All of the above
20.The mass defect for an isotope was found to be 0.410 amu/atom. Calculate the binding energy in kJ/mol of atoms. (1 J = 1 kgm2/s2) 3.69 x 1010 kJ/mol 1.23 x 1020 kJ/mol 3.69 x 1013 kJ/mol 1.23 x 103 kJ/mol





M. V. Muthiah Government Arts College For Women (Affiliated to Mother Teresa Women's University, Kodaikanal) Re-accredited with 'A' Grade by NAAC Dindigul-624 001, TamilNadu

CERTIFICATE OF APPRECIATION (Certificate ID:CNPXI0-CE000447)

ANIL KUMAR MAURYA BA 2nd year Banaras Hindu University has successfully completed the National Level E-Quiz on Chemistry conducted by the Department of Chemistry, M.V. Muthiah Government Arts College, Dindigul-624001, Tamil Nadu on 03/06/2020 & 04/06/2020.

M. okanalakukmi Dr.M.Dhanalakshmi Convener

Dr.C.Ladha Pooranam Principal